

VA Application Note No. V - 78

Title:	Determination of antimony in zinc plant electrolytes (concentrated ZnSO₄ solutions)
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Summary:	The concentration of total Sb in zinc plant electrolytes is determined by anodic stripping voltammetry (ASV) in 5 mol/L HCl. If 0.6 mol/L HCl is used only the concentration of antimony(III) is determined selectively. The interference of an excess of Cu is suppressed by the selective oxidation of Cu. Nevertheless the concentration of Cu in the sample limits the amount of sample that can be used for the determination.
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Sample:	Zinc plant electrolyte (concentrated ZnSO ₄ solution)
Sample preparation:	None

Analysis of Sb(total)

HCl solution	w(HCl) = 30%	
Measuring solution	5 mL H ₂ O + 5 mL HCl solution + 0.1 mL zinc plant electrolyte	
Working electrode (WE)	MME (Multi Mode Electrode)	6.1246.020
Auxiliary electrode (AE)	Pt	6.0343.000
Reference electrode (RE)	Reference system: Ag/AgCl/KCl (3 mol/L)	6.0728.020
	Intermediate electrolyte: c(KCl) = 3 mol/L	6.1245.010

Parameters

Working electrode	HMDE
Stirrer speed	2000 rpm
Mode	DP
Purge time	300 s
Deposition potential 1	-0.4 V
Deposition time 1	180 s
Deposition potential 2 (selective oxidation)	-0.27 V
Deposition time 2 (selective oxidation)	10 s
Equilibration time	10 s
Pulse amplitude	0.01 V
Start potential	-0.4 V
End potential	-0.1 V

	Voltage step	0.004 V
	Voltage step time	0.1 s
	Sweep rate	0.04 V/s
	Peak potential Sb	-0.18 V

Analysis of Sb(III)

HCl solution w(HCl) = 30%

Measuring solution 10 mL H₂O
+ 0.6 mL HCl solution
+ 1 mL zinc plant electrolyte

Working electrode (WE) MME (Multi Mode Electrode) 6.1246.020

Auxiliary electrode (AE) Pt 6.0343.000

Reference electrode (RE) Reference system: Ag/AgCl/KCl (3 mol/L) 6.0728.020

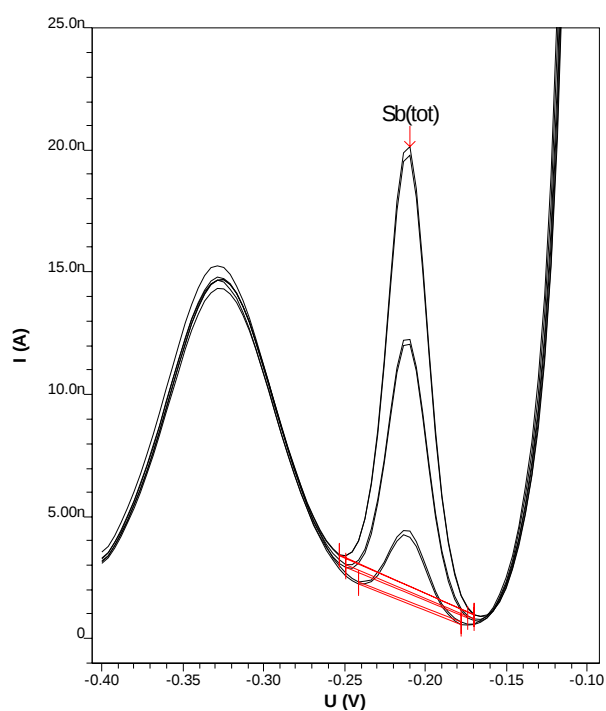
Intermediate electrolyte: c(KCl) = 3 mol/L 6.1245.010

Parameters

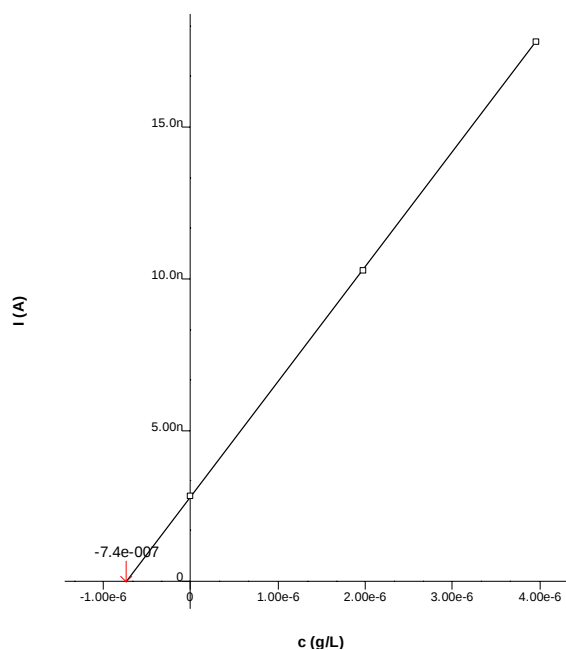
Working electrode	HMDE
Stirrer speed	2000 rpm
Mode	DP
Purge time	300 s
Deposition potential 1	-0.4 V
Deposition time 1	120 s
Deposition potential 2 (selective oxidation)	-0.2 V
Deposition time 2 (selective oxidation)	60 s
Equilibration time	10 s
Pulse amplitude	0.01 V
Start potential	-0.2 V
End potential	-0.05 V
Voltage step	0.004 V
Voltage step time	0.1 s
Sweep rate	0.04 V/s
Peak potential Sb	-0.13 V

Results:	Sb(total)	Sb(III)
	74 µg/L	21 µg/L

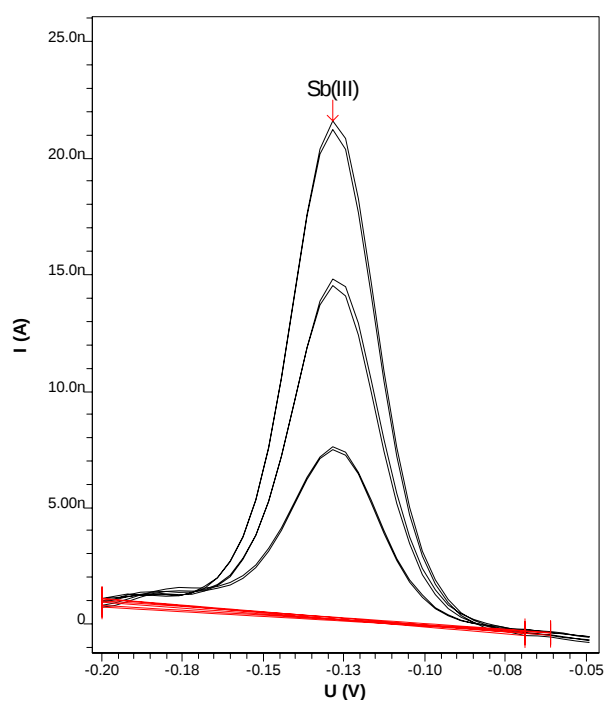
Determination of Sb(total)



Sb(tot)
c = 74.377 µg/L
+/- 1.938 µg/L (2.61%)



Determination of Sb(III)



Sb(III)
c = 21.322 µg/L
+/- 0.320 µg/L (1.50%)

